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Silberstein, 2003 Element Subscript Oxidation state Sum of oxidation states K Cr +2 TOTAL -14 9. 10. 11. [al potassium is a group one metal; its oxidation state is always +1 [bl oxygen usually has an oxidation state of—2 [CJ the sum of oxidation states of each element is the product of the subscript and the oxidation state

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Read the poem, then answer the questions. Name: _____ Follow the directions on the back after you finish the front. Assignment : _____ Period: _____ Elements, Compounds, and Mixtures By Evan P. Silberstein Elements are simple things That can't be broken down. When put together

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Evan Silberstein Chemistry Answer Key

Write answers in the boxes below. 23. What is the mass of 0.5 mole of ammonia, NH₃? 24. What is the mass of 2 moles of ? Ca (OH)_z + 3242 25. What is the mass of 4.2 moles of H₃P₀₄? 26. What is the mass Of 0.75 mole of butane, C₄H₈ 911,6 47 5Ø 27. What is the mass of 2 moles of gypsum, CaC₁₂.2H₂O ? 2HzO 2(35) 4

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